

Exam Settings

Last Modified on 08/01/2024 1:24 pm EDT

Smartwork allows you to assign exams with the Exam Mode setting.

Hide All Answers

Exam Mode

In the assignment editor, you can toggle Exam Mode ON and OFF. By setting Exam Mode ON, the exam settings will be applied to the assignment.

Settings

11868 (2020 Fall Semester)

Assignment Name: Final Exam

Description:

Grades Accepted Until (GAU): 12/04/2020 11:59 PM (GMT-05:00) Eastern Time

Late work accepted days after the GAU, at % penalty per day.

Exam Mode: OFF Exam settings

SHOW ADDITIONAL SETTINGS

Adaptive: OFF These are the settings for the Knewton adaptive portion of this assignment.

Questions ADD QUESTIONS

Estimated total time: 49 min Total points: 28 pts

Questions	Question Type	Avg.Time	Points	Attempts	Grade Penalties	Hints	
Adjust all question settings:							
01 Equilibrium and Free Energy 2...	Mixed	03:08	4	∞	0%	ON	✎ ✕
02 Identify Relationships When R...	Mixed	03:16	2	∞	0%	ON	✎ ✕
03 Silver and Tin Voltaic Cell	Chemical Equation	07:06	3	∞	0%	ON	✎ ✕
04 Cell Potential, Free Energy, an...	Mixed	09:57	3	∞	0%	ON	✎ ✕
05 [NEW] Determine Ecell, anode...	Mixed	--:--	3	∞	0%	ON	✎ ✕
06 Complete and Balance Nuclea...	Chemical Equation	06:41	1	∞	0%	ON	✎ ✕
07 Balancing Nuclear Equations	Chemical Equation	05:16	4	∞	0%	ON	✎ ✕

PREVIEW Not published PUBLISH SAVE

Exam Settings

<p>Assignment Name:</p> <input type="text" value="Exam"/>	<p>Description:</p> <input type="text"/>
<p>Grades Accepted Until (GAU):</p> <p>08/14/2023 11:59 PM (GMT-04:00) Eastern Time</p> <p><input type="checkbox"/> Late work accepted <input type="text"/> days after the GAU, at <input type="text"/> % penalty per day.</p>	<p>Exam Mode: ON <i>Exam settings</i></p>
<p>Time Limit: OFF</p> <p><input type="text"/> minutes</p>	<p>Ungraded Practice: <i>You cannot change this setting in Exam Mode</i></p> <p>Exam mode: never allow ungraded practice</p>
<p>Show Solution: <i>You cannot change this setting in Exam Mode</i></p> <p>Exam mode: never show solution</p>	<p>Show Feedback: <i>You cannot change this setting in Exam Mode</i></p> <p>Exam mode: never show feedback and correct/incorrect</p>
<p>Show Ebook Link: ON</p>	<p>Show Periodic Table: OFF</p>
<p>Randomize Questions: OFF</p>	<p>Show Student Score In:</p> <p>Percentages</p>
<p>Show Question Title Text: <i>Learn More</i></p> <p>Show the question title text to students</p>	

HIDE ADDITIONAL SETTINGS ▲

When Exam Mode toggled ON, the following assignment settings will be automatically defaulted and **cannot** be changed:

- **Show Solution** will be defaulted to “never show solution” during the assignment. Solutions will be shown after the Grades Accepted Until date or after the Late Work Penalty date have passed.
- **Ungraded Practice** will be defaulted to “never allow ungraded practice”
- **Show Feedback** will be defaulted to “never show feedback and correct/incorrect alerts”
- **Attempts** will be defaulted to 1 attempt
- **Grade Penalties** will be defaulted to 0% grade penalty
- **Adaptive** will be set to OFF

If you would like to change the settings that are defaulted for the exam settings, please toggle Exam Mode to OFF to apply those settings to a regular assignment.

Grades Accepted Until Date

The Grades Accepted Until date can only be updated and saved prior to the expiration date applied to the Exam assignment. If the Grades Accepted Until date expires, you will see the following pop-up and will not be able to update the date.

⚠ Unable to edit GAU
✕

Once the Grades Accepted Until (GAU) date has passed, students have access to the solutions. Therefore, extending the GAU date for this exam is not permitted.

Adaptive (Chemistry)

Adaptive settings cannot be used when Exam Mode is ON. If you would like to change the settings to use Adaptive, please toggle Exam Mode to OFF to create an Adaptive assignment.

Student Exceptions

If Exam Mode is on, some student exceptions can be applied.

Student Detail [X]

Violet Beauregard's results [EDIT] [EXPORT]

Publish: **ON** [III]

Grades Accepted Until (GAU): *You cannot change this setting in Exam Mode*
12/04/2020 [📅] 11:59 PM [▼] (GMT-05:00) Eastern Tin [▼]

Time Limit: **OFF**
[] minutes

Ungraded Practice: *You cannot change this setting in Exam Mode*
Exam mode: never allow ungraded practice [▼]

You cannot edit Attempts in Exam Mode

Question	Score	Attempts	Submission Date	Time Spent	Reset
01. Equilibrium and Free...	4/4pts	1/ 1 [▼]	10/26/20 - 01:05:39PM	00:00:22	[🔄]
02. Identify Relationship...	1/2pts	1/ 1 [▼]	10/26/20 - 01:04:34PM	00:09:28	[🔄]
03. Silver and Tin Voltaic...	3/3pts	1/ 1 [▼]	10/26/20 - 01:14:32PM	00:06:44	[🔄]
04. Cell Potential, Free E...	1/3pts	1/ 1 [▼]	10/26/20 - 01:16:17PM	00:01:41	[🔄]
05. [NEW] Determine E...	2/3pts	1/ 1 [▼]	10/26/20 - 01:17:04PM	00:00:44	[🔄]
06. Complete and Balan...	1/1pt	1/ 1 [▼]	10/26/20 - 01:18:43PM	00:01:37	[🔄]
07. Balancing Nuclear E...	4/4pts	1/ 1 [▼]	10/26/20 - 01:20:32PM	00:01:47	[🔄]

[🔄] RESET ALL RESULTS [X] CANCEL [SAVE]

In the Student Detail, the following settings **cannot** be updated when Exam Mode is ON:

- Publish
- Grades Accepted Until date
- Ungraded Practice
- Attempts

However, you will be able to update or use the following settings:

- Time Limit
- Score
- Question Reset
- Reset All Results

Exam Assignment

When a student begins their Exam, they will see that they are limited to one attempt per question.

Smartwork 5 CHEMISTRY: THE SCIENCE IN CONTEXT, 6E vbeauregard@wnorton.edu

Final Exam

Welcome to Smartwork! This assignment is designed with rich feedback to guide you as you learn.

EXAM 12/04/20

Exam is accepted until December 4th, 2020, at 11:59 PM (Eastern Time). Your grades will be visible after this date.

0 OF 12 QUESTIONS COMPLETED ▶ BEGIN EXAM

Question	Type	Points	Attempt	Status
01 Pictured are energy diagrams for three reactions sh...	Mixed	-/4	-/1	Not Started
02 What is the value of G at which a reaction becomes ...	Mixed	-/2	-/1	Not Started
03 A voltaic cell is based on the reduction of Ag aq to A...	Chemical Equation	-/3	-/1	Not Started
04 Watch the ChemTour animation below on cell poten...	Mixed	-/3	-/1	Not Started
05 A voltaic cell consists of a standard hydrogen electr...	Mixed	-/3	-/1	Not Started
06 Several isotopes of curium can be synthesized by bo...	Chemical Equation	-/1	-/1	Not Started
07 Complete the following nuclear reactions used in th...	Chemical Equation	-/4	-/1	Not Started
08 In the summer of 2003 a team of American and Rus...	Chemical Equation	-/3	-/1	Not Started
09 Amino acids found in proteins are classified as alpha...	Sorting	-/1	-/1	Not Started
10 The Fischer projection of the monosaccharide galac...	Molecule Drawing	-/2	-/1	Not Started

Final Exam 12/04/20 EXAM vbeauregard@wnorton.edu

This is a Mixed question / It is worth 2 points / You have 1 attempt / Your grade will be visible after 12/04/2020

02 Question (2 points) See page 870

What is the value of ΔG at which a reaction becomes spontaneous? Express your answer numerically.

1st attempt

Part 1 (1 point) See Periodic Table

kJ/mol

Part 2 (1 point)

Which of the equations below can be used to solve for the temperature at which a reaction becomes spontaneous?

Choose one:

$T = \frac{-(\Delta G + \Delta H)}{\Delta S}$

$T = \frac{-(\Delta G - \Delta H)}{\Delta S}$

0 OF 12 QUESTIONS COMPLETED ◀ 02/12 ▶ ▶ SUBMIT ANSWER

When a student submits their attempt, they will not see a Feedback pop-up or an indicating mark to confirm if the answer was correct or incorrect. Once a student submits their attempt, the question will be marked as QUESTION COMPLETED.

Final Exam 12/04/20 EXAM vbeauregard@wnorton.edu

This is a Mixed question / It is worth 2 points / You have no attempts remaining / Your grade will be visible after 12/04/2020

02 Question (2 points) [See page 870](#)

What is the value of ΔG at which a reaction becomes spontaneous? Express your answer numerically.

1st attempt

Part 1 (1 point) [See Periodic Table](#)

0 kJ/mol

Part 2 (1 point)

Which of the equations below can be used to solve for the temperature at which a reaction becomes spontaneous?

Choose one:

$T = \frac{-(\Delta G + \Delta H)}{\Delta S}$

$T = \frac{-(\Delta G - \Delta H)}{\Delta S}$

2 OF 12 QUESTIONS COMPLETED < 02/12 > QUESTION COMPLETED

If a student returns to the assignment page, they will see their question attempt and question status. However, a student will not see the points they earned on each question or their Exam grade until the Grades Accepted Until date expires. If there is a Late Work Penalty, the grade will be shown after penalty period expires.

Smartwork 5 CHEMISTRY: THE SCIENCE IN CONTEXT, 6E vbeauregard@wnorton.edu

Final Exam EXAM

Welcome to Smartwork! This assignment is designed with rich feedback to guide you as you learn.

12/04/20

Exam is accepted until December 4th, 2020, at 11:59 PM (Eastern Time). Your grades will be visible after this date.

12 OF 12 QUESTIONS COMPLETED [REVIEW EXAM](#)

Question	Type	Points	Attempt	Status
01 Pictured are energy diagrams for three reactions sh...	Mixed	-/4	1/1	Completed
02 What is the value of G at which a reaction becomes ...	Mixed	-/2	1/1	Completed
03 A voltaic cell is based on the reduction of Ag aq to A...	Chemical Equation	-/3	1/1	Completed
04 Watch the ChemTour animation below on cell poten...	Mixed	-/3	1/1	Completed
05 A voltaic cell consists of a standard hydrogen electr...	Mixed	-/3	1/1	Completed
06 Several isotopes of curium can be synthesized by bo...	Chemical Equation	-/1	1/1	Completed
07 Complete the following nuclear reactions used in th...	Chemical Equation	-/4	1/1	Completed
08 In the summer of 2003 a team of American and Rus...	Chemical Equation	-/3	1/1	Completed
09 Amino acids found in proteins are classified as alpha...	Sorting	-/1	1/1	Completed
10 The Fischer projection of the monosaccharide galac...	Molecule Drawing	-/2	1/1	Completed
11 Complete the drawing of the structure of deoxythy...	Molecule Drawing	-/1	1/1	Completed
12 Selenocysteine can exist as two enantiomers stereo...	Molecule Drawing	-/1	1/1	Completed

